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Cheema, Waqas Akram; Kaarsholm, Kamilla Marie Speht; Andersen, Henrik Rasmus

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ORAL 32

DESTRUCTION OF DBP_s AND THEIR PRECURSORS IN SWIMMING POOL WATER BY COMBINED UV-TREATMENT AND OZONATION

Cheema WA^{1,2}, Kaarsholm KMS¹, Andersen HR¹

¹Technical University of Denmark, Bygning Lyngby, Denmark

²National University of Sciences & Technology, Islamabad, Pakistan

Highlights

- UV treatment increased the reactivity of pool water to both chlorine and ozone
- Ozonation of UV-treated water decrease chlorine reactivity
- Genotoxic trichloronitromethane formed by ozonation was removed with UV treatment
- Continuous UV/ozone treatment decreases chlorine by-product formation
- Continuous UV/ozone treatment predicted to improve chlorinated pool water quality

Aims

The aim of the current study was to investigate the effect of a combined treatment system on DBP formation. As both ozone and chlorine preferably react with electrophilic groups in compounds [1], [2], we hypothesise that reactivity to chlorine, created by the UV treatment of dissolved organic matter in pool water, might also mean that there is increased reactivity to ozone and that ozonation might remove the chlorine reactivity created by UV treatment. Therefore, we first performed an experiment to range-find the effect of swimming pool water UV activation on chlorine reactivity. Second, an experiment was carried out to characterise the effect of adding various doses of ozone to pool water, with or without UV pre-treatment, before chlorination to study the effect on chlorine reactivity and the formation of chlorination by-products. Finally, the possible effect on chlorination by-product formation was investigated by a repeated, combined UV-ozone treatment interchanged with chlorination (repeated cycles of UV followed by ozone with subsequent chlorination). Toxicity estimation was used to evaluate water quality.

Methods

- UV treatment
Treatment was conducted using a quasi-collimated beam apparatus with a doped, medium pressure lamp (P = 700 W, ScanResearch, Denmark). To ensure constant spectra and emission output, the lamp was turned on half an hour before the experiment. Petri dishes (350 mL) were used as reaction vessels, while samples were maintained headspace-free and covered by a disc of quartz glass, to limit the volatilisation of the treated sample. To ensure homogeneity during irradiation, samples were mixed gently with a stirrer. The UV dose was determined according to a method described by Hansen et al. [3]. In summary, UV exposure in the collimated beam set-up was correlated to a real flow-through system on a pool, using the removal of combined chlorine. The UV system needs 1.0 kWh/m³ to remove 90% of the combined chlorine. For the collimated beam set-up, required radiation time to remove 90% of the combined chlorine from the pool water was 12.3 mins.
- Ozonation

Ozonation was achieved by adding an amount of ozone stock solution to a water sample which resulted in maximum 10% dilution of the sample and the concentrations were back calculated according to actual dilution. Ozone dosage was determined by adding a sufficient amount of potassium indigotrisulfonate and a phosphate buffer to a separate ultra-pure water sample and measuring the absorbance of the unreacted indigotrisulfonate. A detailed description can be found in Hansen et al. [4].

- Chlorination

The formation of DBPs as a result of chlorination was investigated using a standardised DBP formation assay. The effect of chlorine concentration in the assay was also recently investigated by Hansen et al. [5]. In the current study, the same approach was used to simulate chlorination in the pool after the return of UV/ozone-treated water. Water samples were transferred to 40 mL glass vials after treatment in which chlorine and boric acid were added based on the chlorine consumption determined in pre-experimental tests. The aim was to have 1 ± 0.3 mg Cl_2/L after 24 hours at 25°C (measured by ABTS). Chlorination was performed in quintuplicate, with three samples used for DBP analysis and two for the determination of residual chlorine. Samples for DBP analyses were dosed with ammonium chloride solution (50 mg/L), to quench free chlorine which neither affects the already formed DBP [6] nor increases N-DBP formation [7]. The samples were analysed the same day.

Results

We found that UV treatment makes pool water highly reactive to ozone. The created reactivity towards chlorine decreases dose dependently with ozone dosage prior to contact with chlorine. Furthermore, the kinetics of ozone in UV treated pool water changed significantly from a half-life of 5 min to complete consumption in less than 2 min. Ozonation of UV treated pool water induced formation of some DBPs that are not commonly reported in pool water where trichloronitromethane is noteworthy for its genotoxic. However, this created genotoxicity was removed by UV treatments when repeated combined UV/ozone treatment interchanged with chlorination for 24h were conducted. The discovered reaction can form the basis for a new treatment method for swimming pools. The schematic of the proposed system has been shown in figure-1.

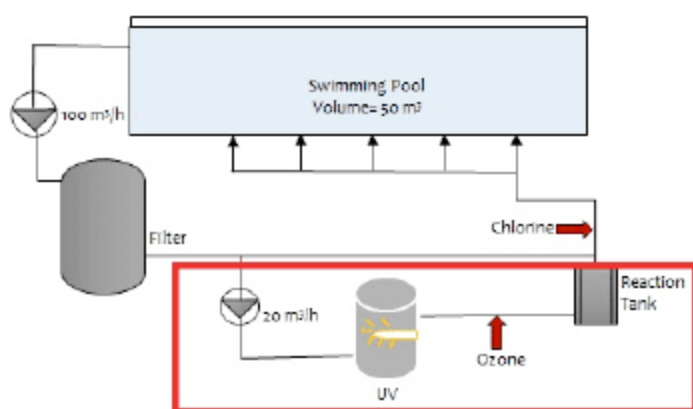


Figure 1: Schematic of proposed system

Conclusions

The treatment of swimming pool water by means of UV irradiation increased chlorine demand. Furthermore, the ozonation of pre-treated UV-irradiated pool water subsequently removed chlorine demand and decreased DBP formation. Combined treatment effectively reduced the level of disinfection by-products in pool water except for trichloronitromethane where an increase was observed. Trichloronitromethane was reduced after repeated treatment cycles and thus UV/ozone treatment is predicted to improve swimming pool water quality.

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Combined UV-treatment followed by ozonation for the removal of by-product precursors in swimming pool water

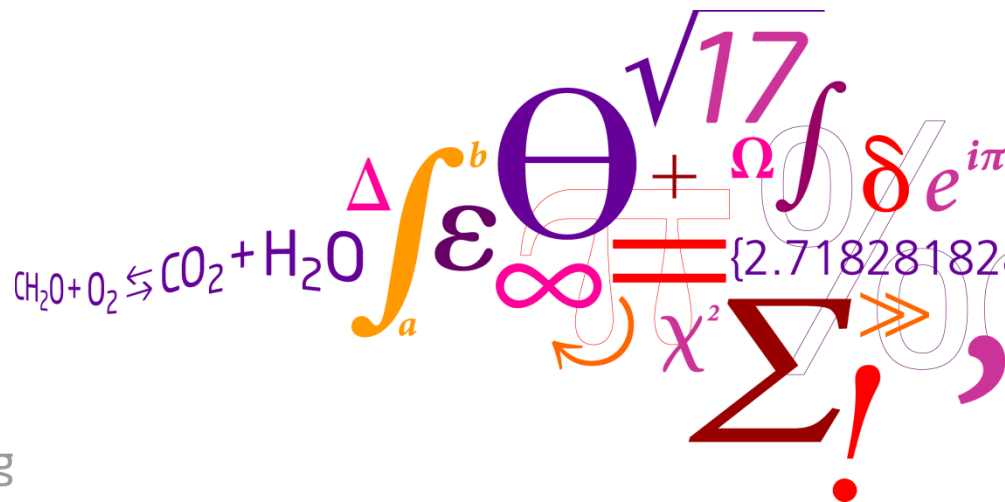
Waqas A Cheema ^{1,2}, Kamilla M.S. Kaarsholm ¹, Henrik R. Andersen ¹

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²National University of Sciences & Technology, Pakistan

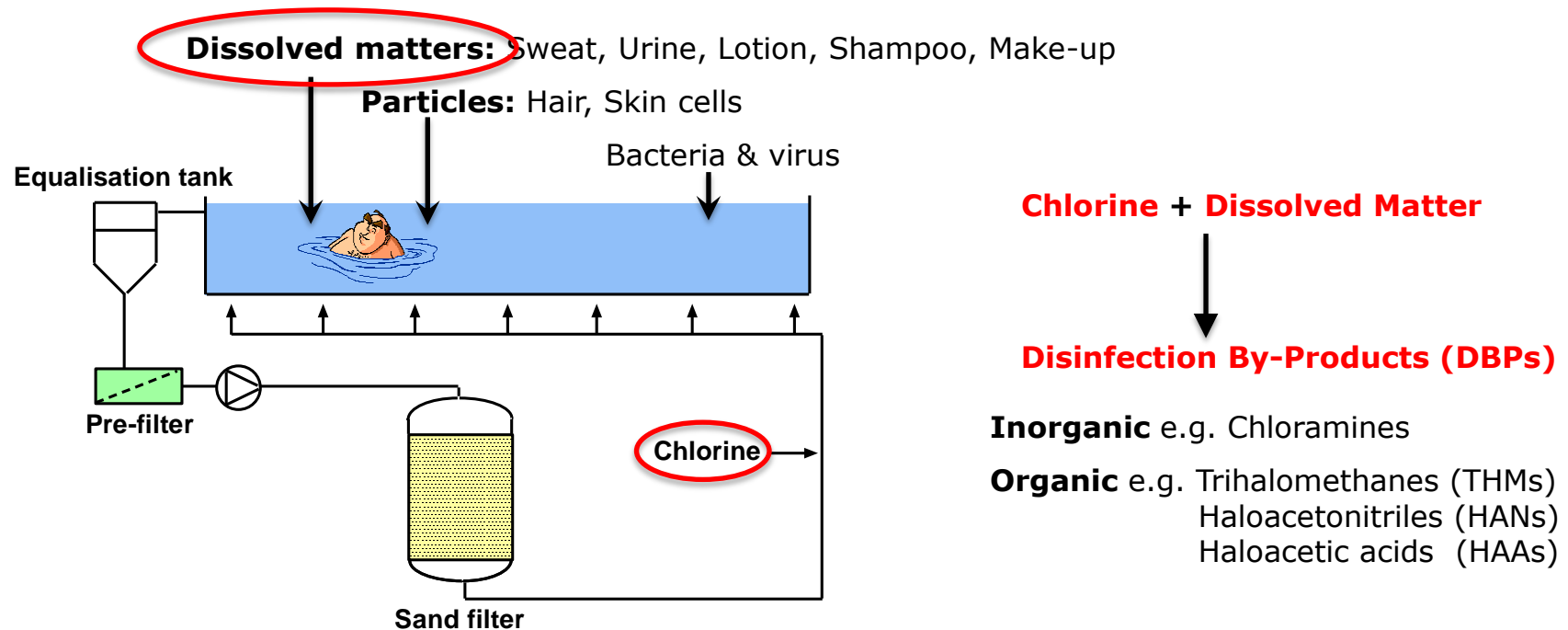
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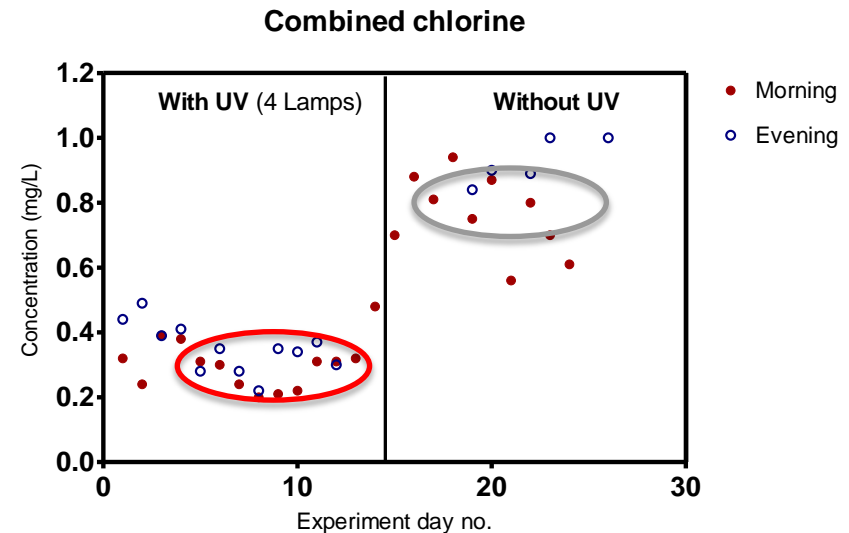
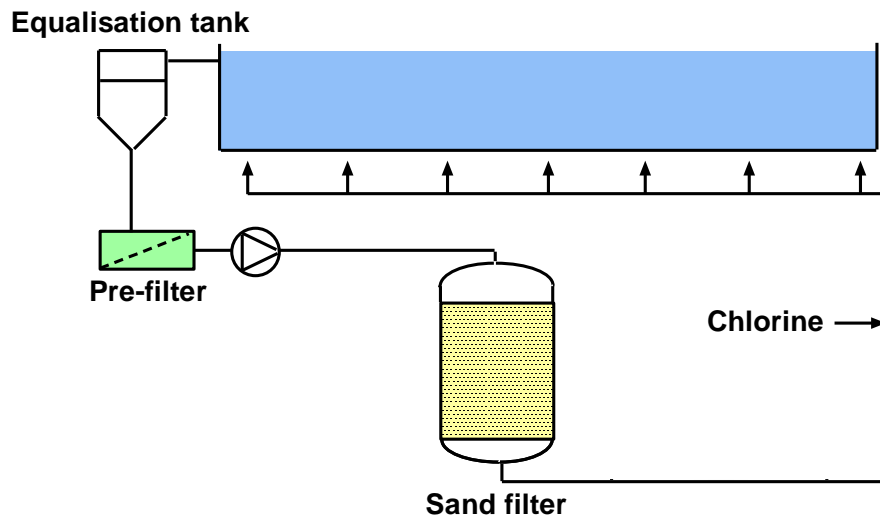
Swimming Pool

Disinfection By-Products



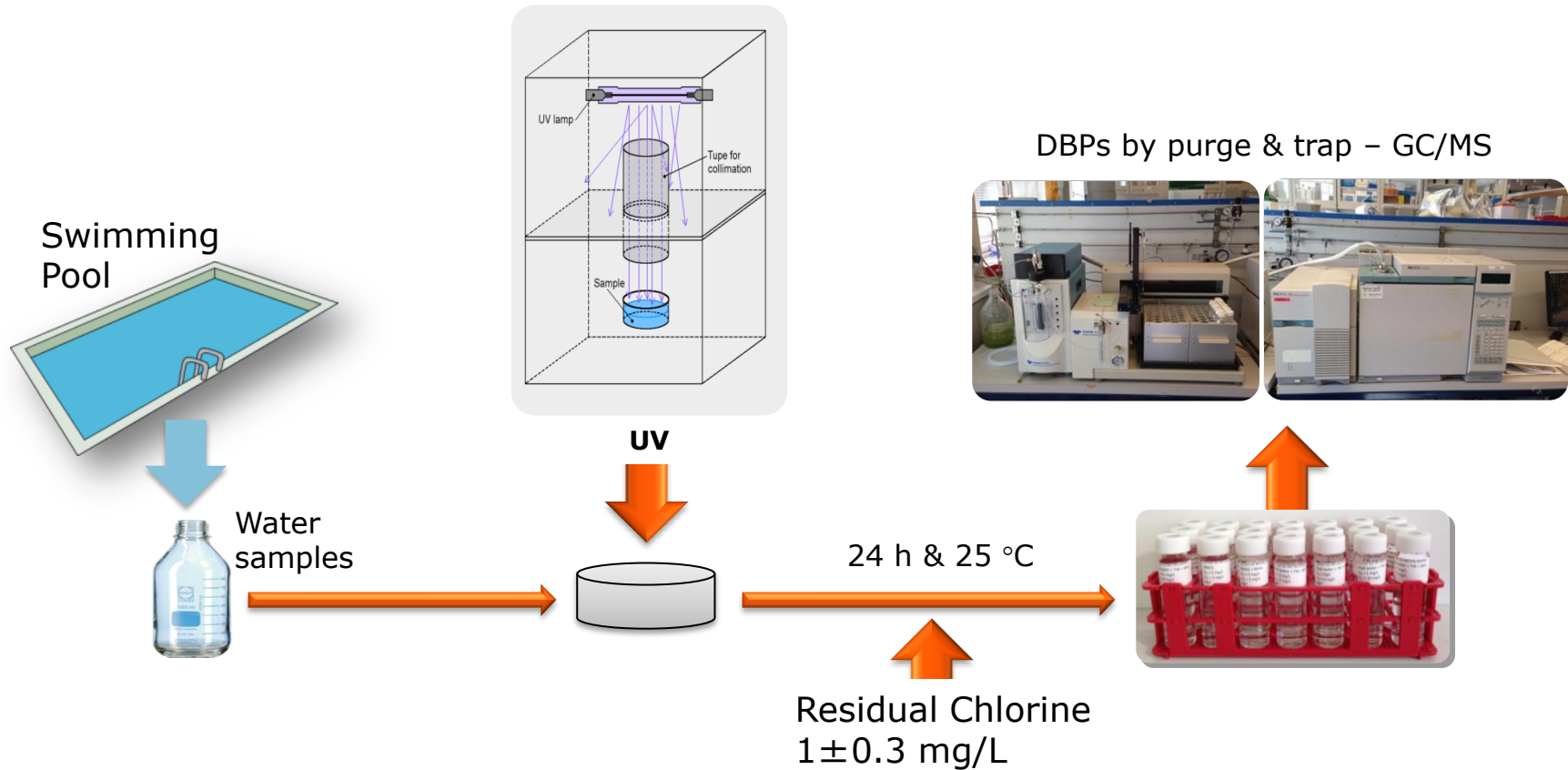
UV Treatment

Gladsaxe pool Copenhagen-Denmark

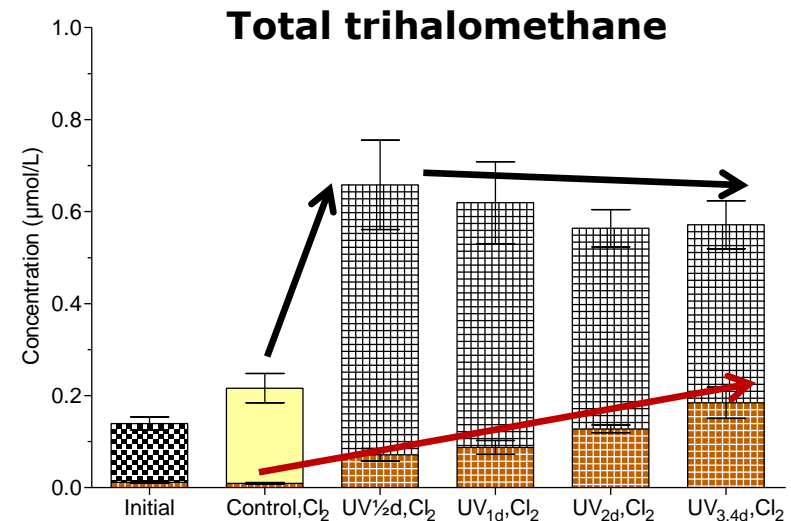
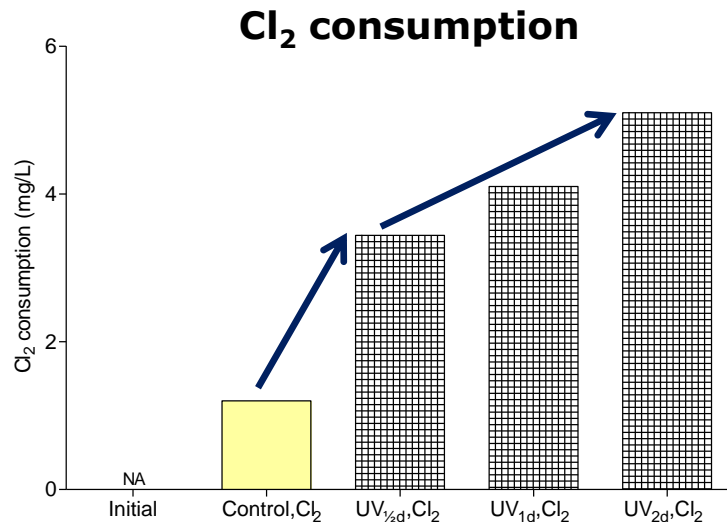


- UV treatment followed by Cl_2 → decreased combined Cl_2

UV Treatment Experimental Setup



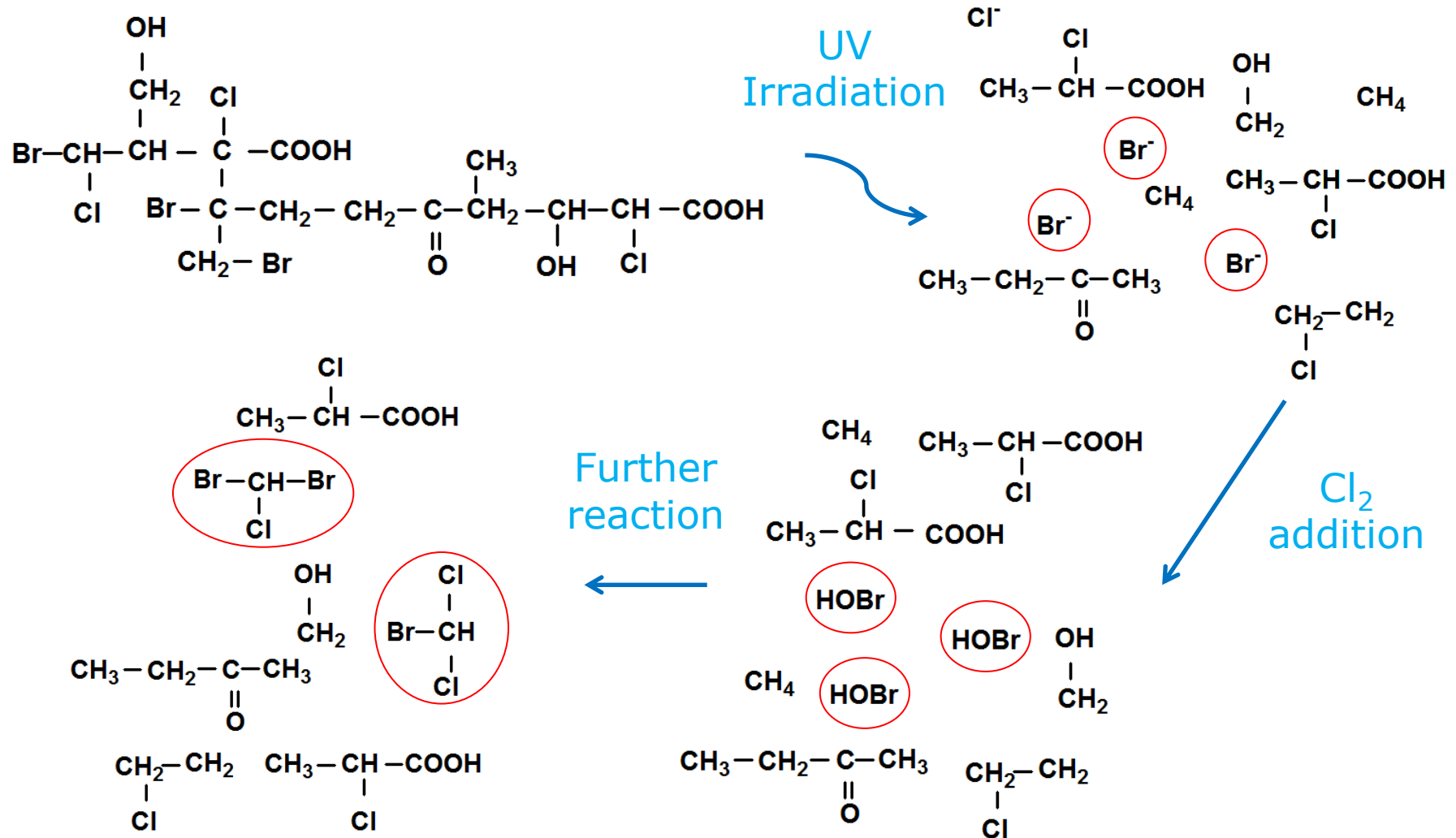
UV Treatment Results



- UV treatment followed by Cl₂ → increased Cl₂ reactivity
- Increasing UV dose followed by Cl₂ → no effect on Total THMs
- Increasing UV dose followed by Cl₂ → increased Br-THMs

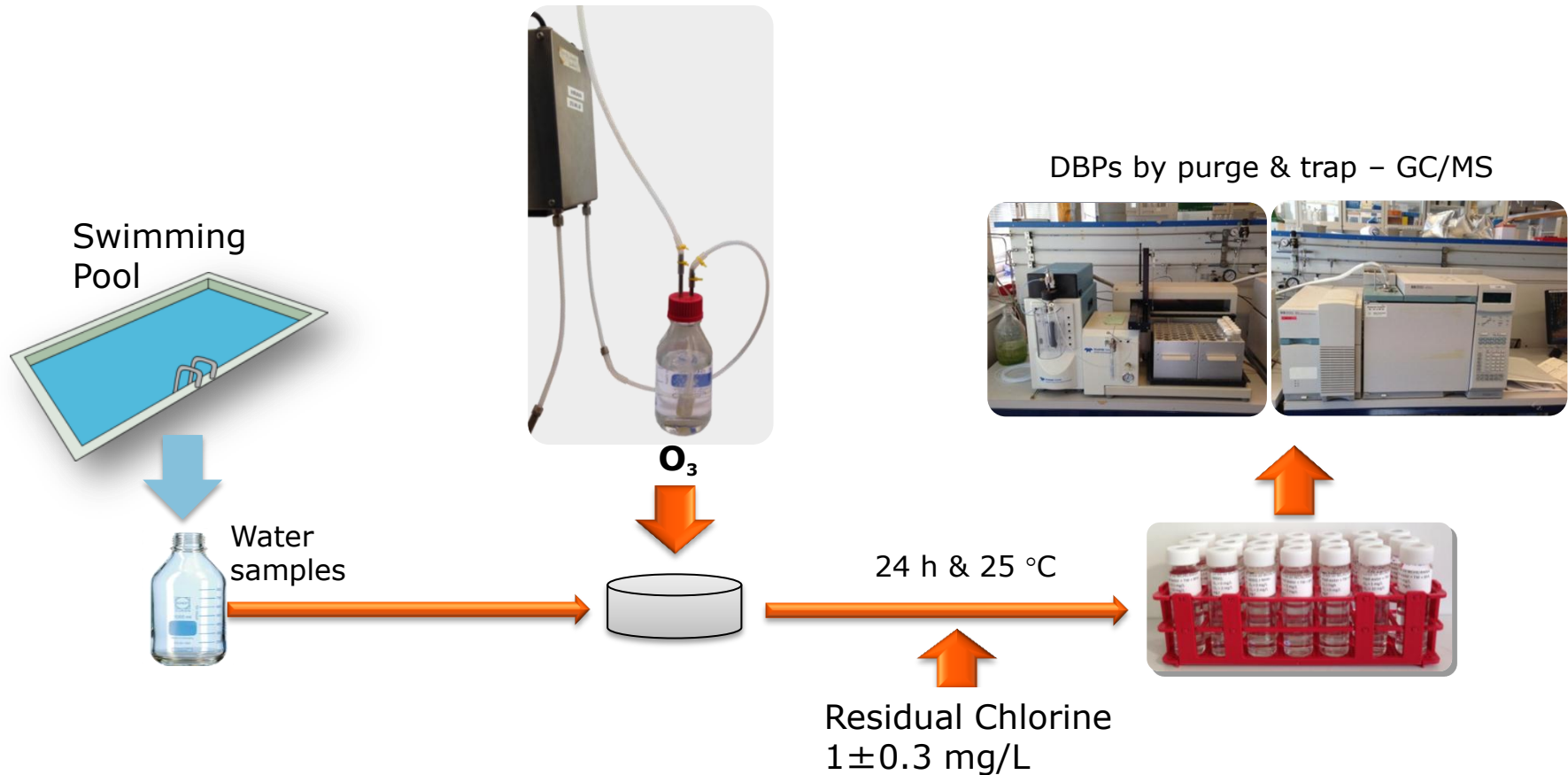
UV Treatment

Br-Cl-DBP Formation Theory



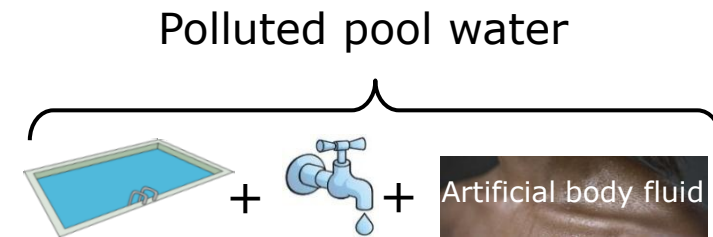
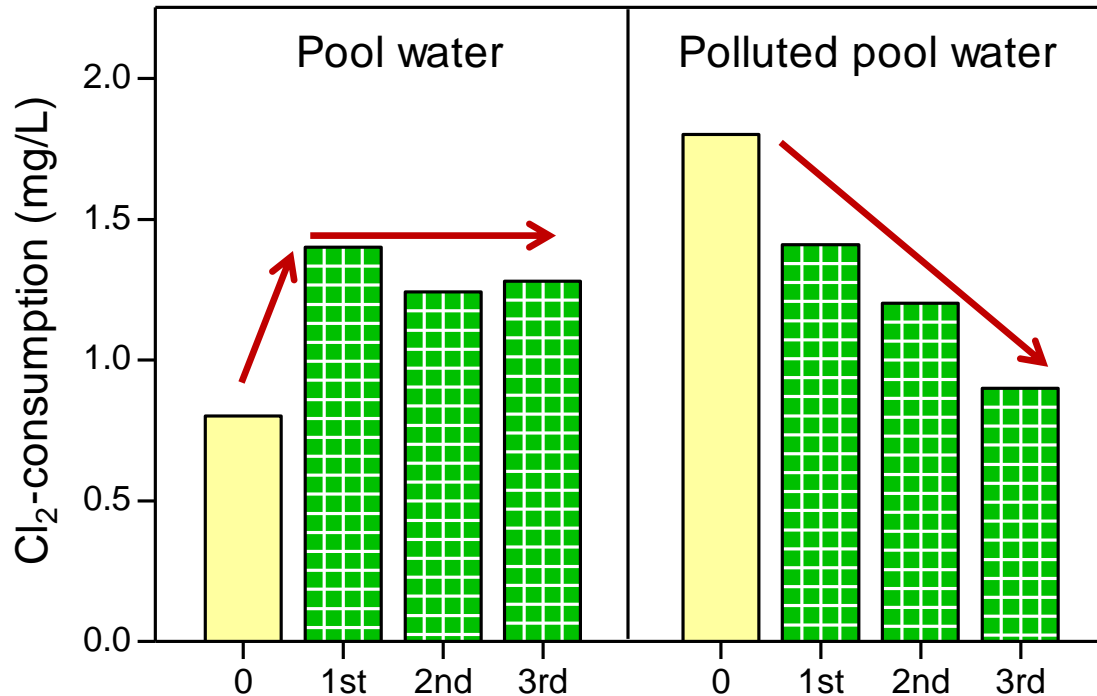
Ozonation

Experimental setup



Ozonation

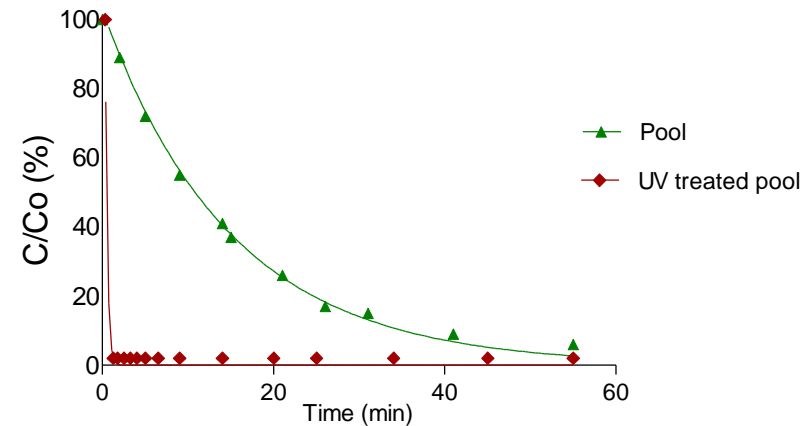
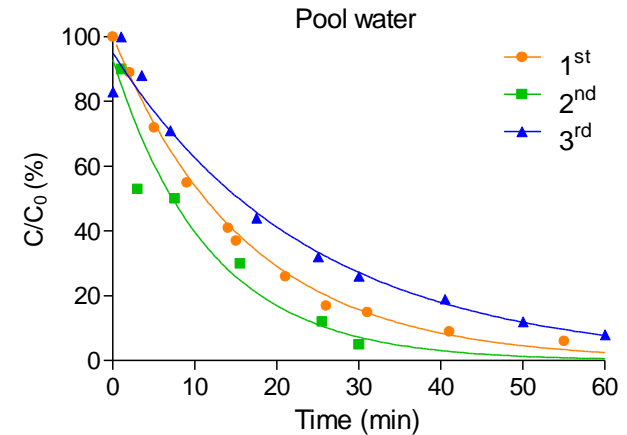
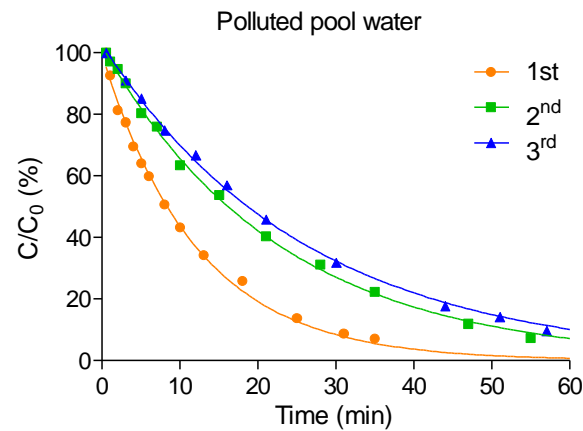
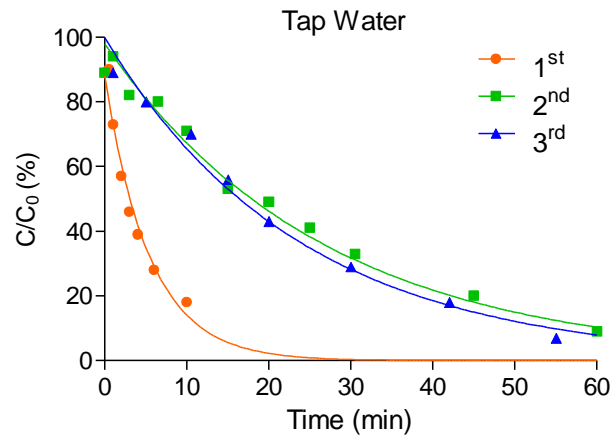
Chlorine consumption



- Ozonation of pool water → increased Cl_2 reactivity
- Ozonation of polluted pool water → decreased Cl_2 reactivity

Ozonation

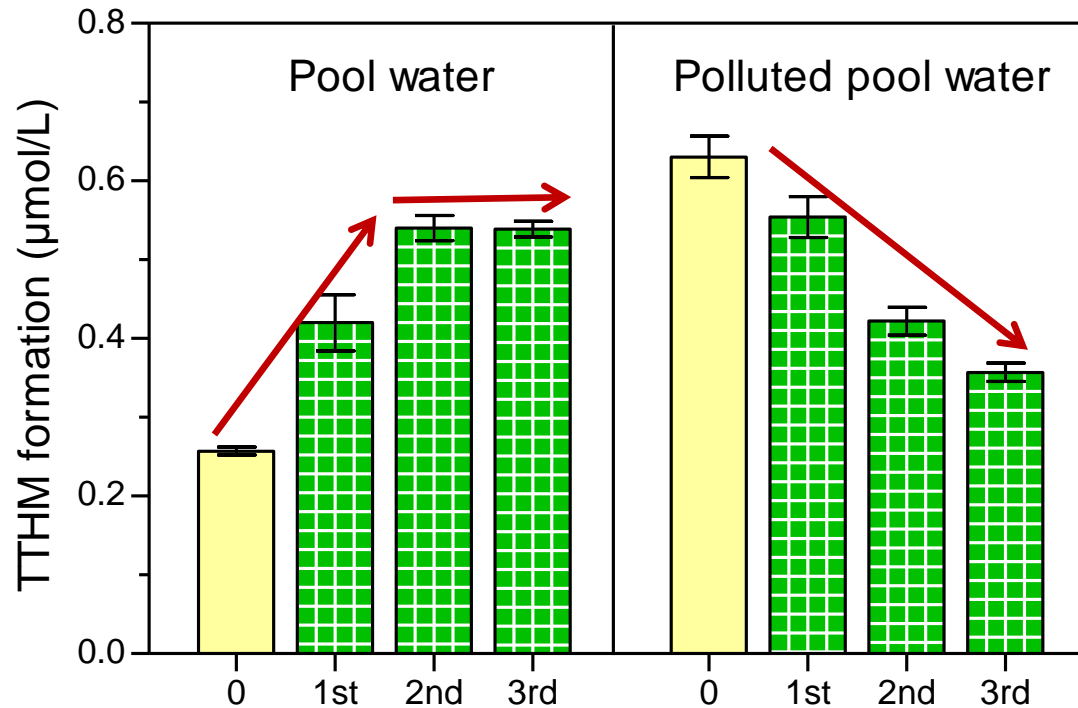
Ozone lifetime



- Slow consumption of ozone
- Fast consumption of ozone

Ozonation

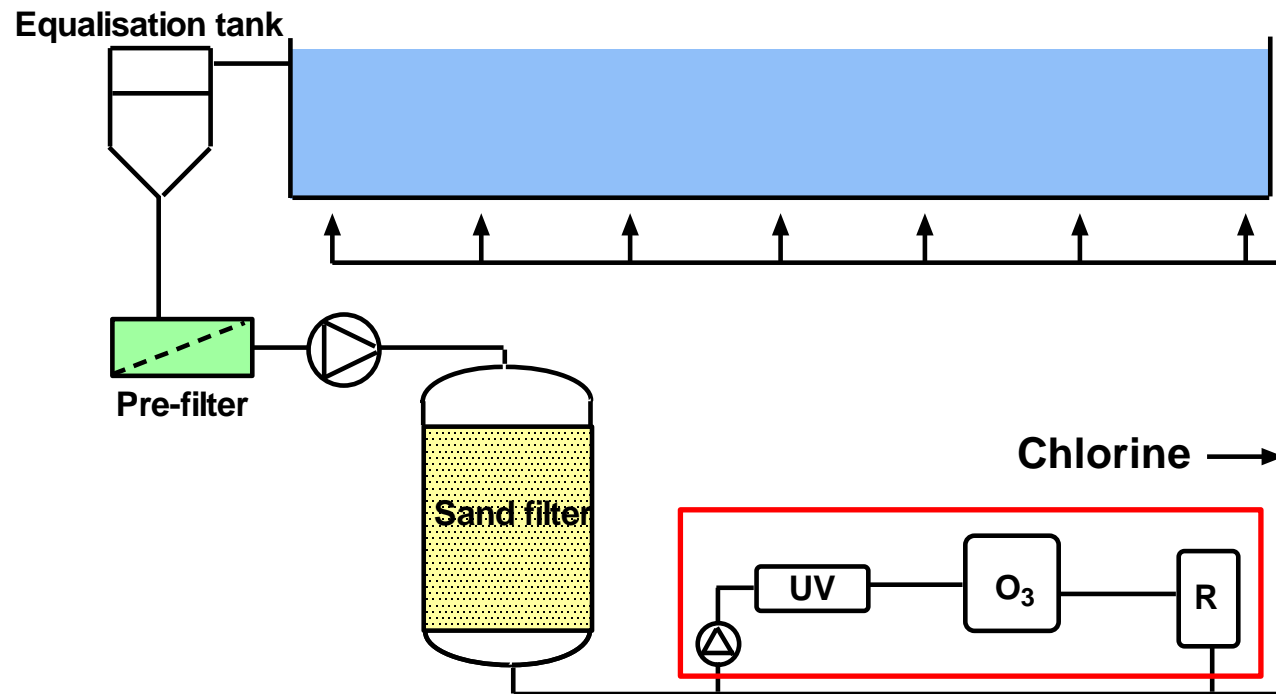
Formation of total trihalomethane



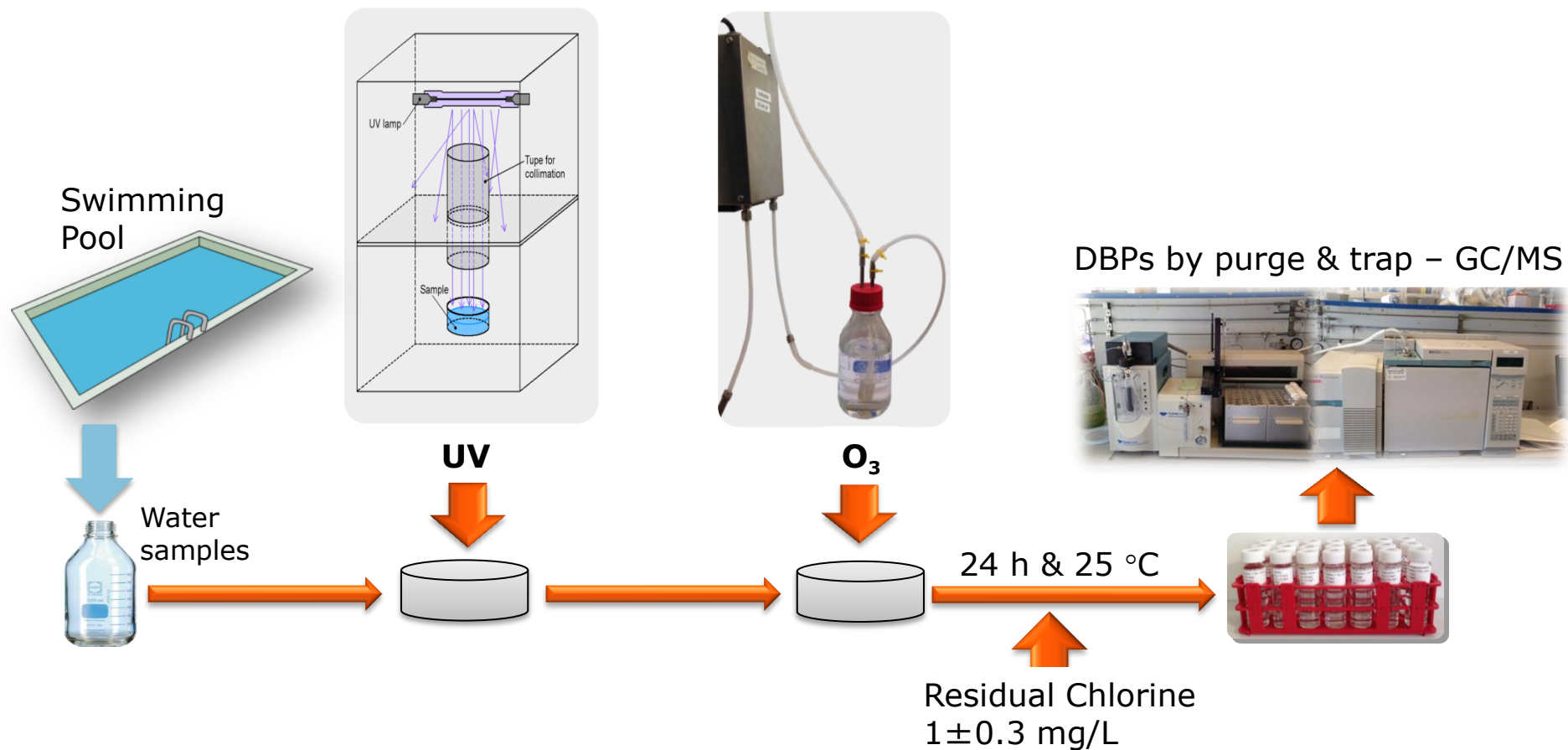
- Ozonation of pool water → increased THM
- Ozonation of polluted pool water → decreased THM

Swimming pools design

Proposed system

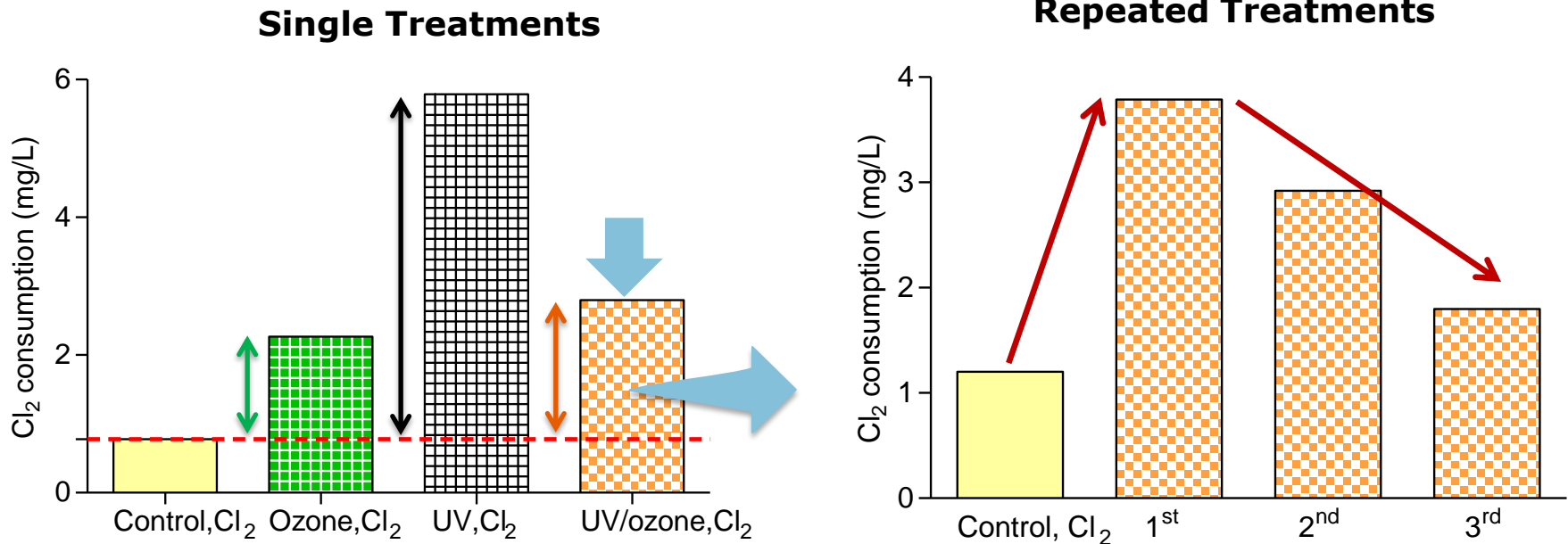


Combined Treatment (UV/O₃) Experimental Setup



Combined Treatment (UV/O₃)

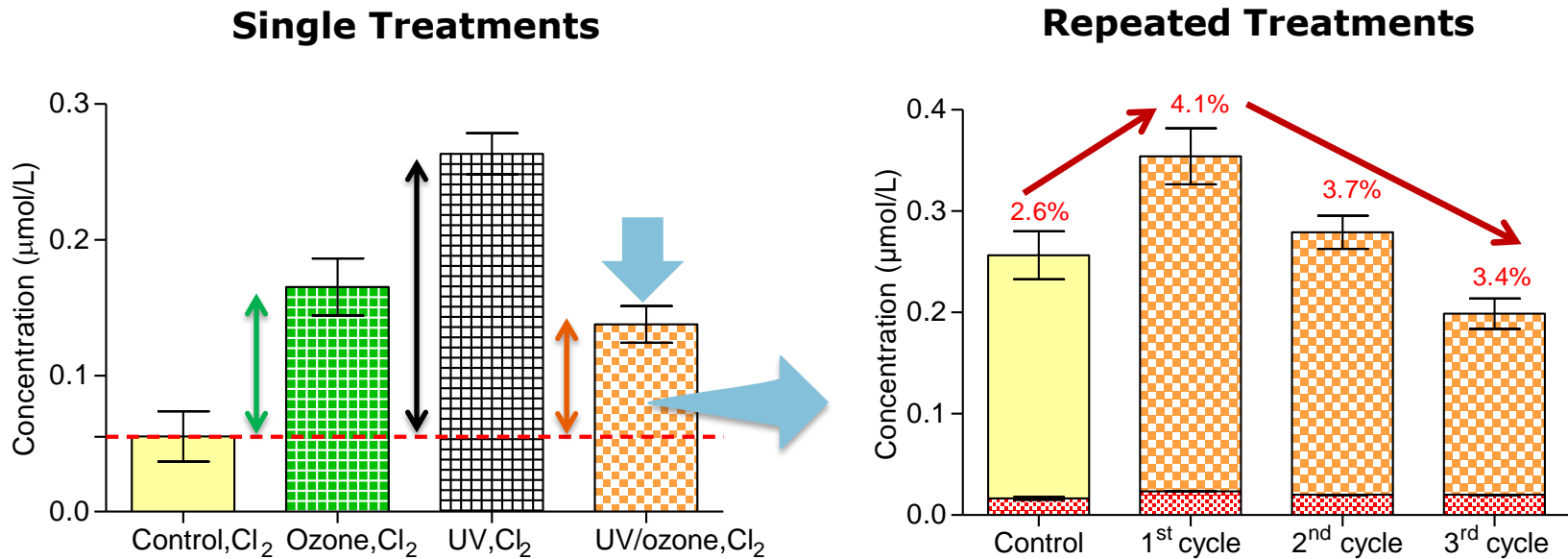
Cl₂ consumption



- Combined treatment followed by chlorination → increased Cl₂ reactivity
- Repeated combined treatments → decreased Cl₂ reactivity

Combined Treatment (UV/O₃)

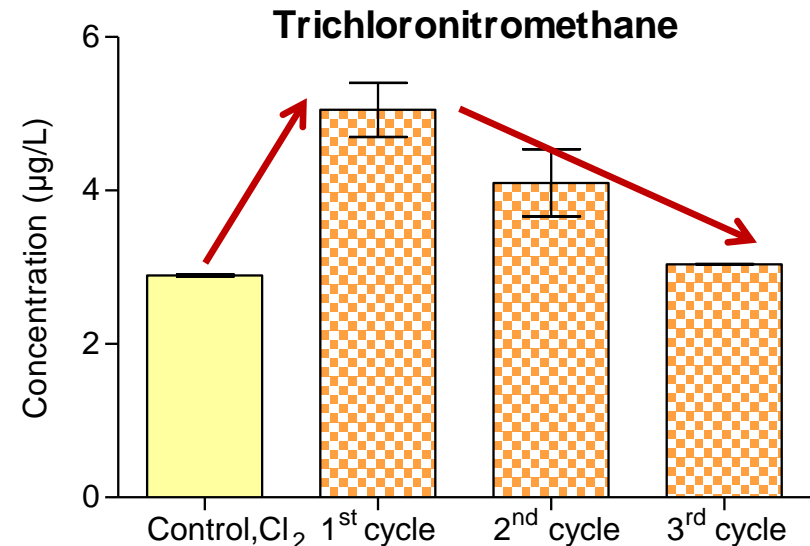
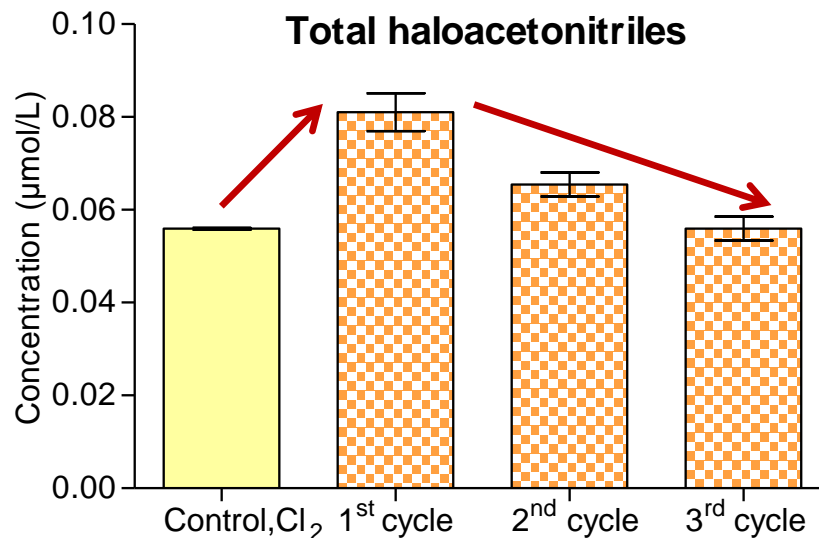
Total trihalomethane formation



- Combined treatment followed by Cl₂ → increased THM
- Repeated combined treatments followed by Cl₂ → decreased THM

Combined Treatment (UV/O₃)

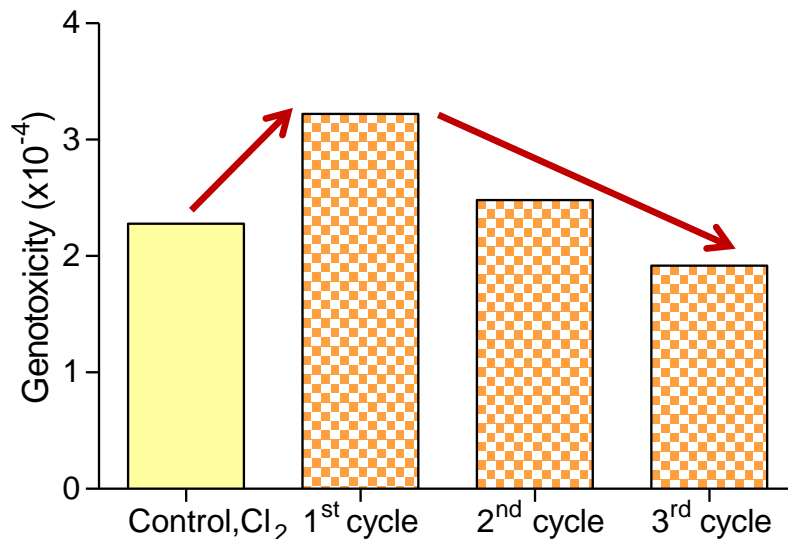
Total HANs & trichloronitromethane



- Repeated (UV/O₃) treatment followed by Cl₂ → decreased Total HAN
- Repeated (UV/O₃) treatment followed by Cl₂ → decreased trichloronitromethane

Combined Treatment (UV/O₃)

Predicted toxicity



$$Toxicity = \sum \frac{C_i}{EC_{50,i}}$$

EC₅₀ taken from Plewa et al. 2008

- The toxicity of the different groups
Haloacetonitriles (HANs) > Haloacetic acids (HAAs) > Trihalomethanes (THMs)
- Repeated (UV/O₃) treatment followed by Cl₂ → decreased toxicity

Thanks for your attention!